TOTAL SYNTHESIS OF β -LACTAM ANTIBIOTICS I. α -THIOFORMAMIDO-DIETHYLPHOSPHONOACETATES

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Since the structure elucidation of cephalosporin C in 1961, ¹ the total synthesis of cephalosporins ² has been the object of considerable interest. The preparation of a series of semi-synthetic 7α -methoxy substituted cephalosporins ³ possessing especially desirable antimicrobial properties has stimulated our efforts toward total synthesis. For commercial reasons, such a synthetic effort must be both unique and simple. The present series of papers describes a practical total synthesis of cephalosporins and 7α -methoxy substituted cephalosporins. This paper describes the synthesis of α -thioformamido-diethylphosphonoacetates (I), key intermediates for further elaboration to 6-H cephalosporins (II). ⁴

1,3,5-Tribenzylhexahydro-s-triazine⁵ reacted with 3 equivalents of diethyl-phosphite at 100° for 6 hours to afford N-benzyl aminomethyldiethylphosphonate, conveniently isolated from Et₂O as its hydrochloride salt 1: 68%; mp 85-87°; nmr (D₂O) τ 8.58 (t, J = 7Hz, CH₃), 6.42 (d, J_{HP} = 14 Hz, CH₂P), 5.72 (d of q, J_{HP} = 8Hz and J = 7Hz, CH₂CH₃), 5.24 (s, CH₂Ph), and 2.43 (s, ArH). Hydrogenolytic debenzylation of 1 over 10% Pd-C in EtOH followed by neutralization with NH₃ in CHCl₃ gave aminomethyl-diethylphosphonate (2) in near quantitative yield: ir (CCl₄) 8.06, 9.41, and 9.69 μ ; nmr (CDCl₃) τ 8.67 (t, J = 7Hz, CH₃), 7.03 (d, J_{HP} = 11 Hz, CH₂P), and 5.86 (p, J = 7Hz,

CH₂CH₃). The Schiff base $\underline{3}$ was prepared in 96-100% yield by stirring amine $\underline{2}$ with benzaldehyde at 0° followed by azeotropic removal of the water formed: ir (CCl₄) 6.09, 8.00, 9.42, and 9.64 μ ; nmr (CDCl₃) τ 5.93 (d of d, J_{HP} = 17.5 Hz and J = 1Hz, CH₂P) and 1.75 (t of d, J = 1Hz and J_{HP} = 5Hz, N=CH).

The Schiff base anion, prepared from $\underline{3}$ and one equivalent of PhLi in THF at -78° , was treated with methyl chloroformate and the reaction mixture was allowed slowly to warm to 0° . Chromatographic purification of the reaction mixture on silica gel gave recovered $\underline{3}$ and acylated Schiff base $\underline{4}$ in 37-43% yield: ir (CCl₄) 5.71, 6.10, 7.94, 9.46, and 9.72 μ ; nmr (CDCl₃) $_{7}$ 6.22 (s, OCH₃), 5.25 (d, J_{HP} = 21 Hz, CHP) and 1.60 (d, J_{HP} = 5Hz, N=CH). The benzylidene group was removed by exchange with 2,4-DNPH·TsOH in BtOH, or better, with p-TsOH·H₂O in Et₂O. Neutralization of the resulting, gummy p-toluenesulfonate salt with K₂HPO₄ gave methyl α -amino-diethylphosphonoacetate ($\underline{5a}$) in 73% yield: ir (CCl₄) 2.94, 5.72, 7.96, 9.48, and 9.70 μ ; nmr (CDCl₃) $_{7}$ 8.17 (br s, NH₂), 6.20 (s, OCH₃), and 6.06 (d, J_{HP} = 20 Hz, CHP). By employing the same procedures, esters 5b-5e were prepared from the corresponding chloroformates.

Thioformylation of amino ester $\underline{5a}$ with ethyl thionoformate⁶ in CCl₄ proceeded smoothly, affording thioformamide $\underline{6a}$ in 69% yield: ir (CCl₄) 3.13, 5.71, 7.01, 8.07, and 9.70 μ ; nmr (CDCl₃) τ 6.18 (s, OCH₃), 3.96 (d of d, J_{HP} = 22 Hz and J = 8.5 Hz, CHP), 0.50 (d, J = 6Hz, S=CH), and 0.27 (br m, NH); uv (EtOH) 265 (e 13,460)m μ . Benzyl ester $\underline{6b}$ was obtained analogously. The remaining esters, $\underline{5c}$ - $\underline{5e}$, behaved differently when subjected to the above thioformylating conditions. For example, amino ester $\underline{5e}$ gave amidine $\underline{7a}$, and $\underline{5c}$ afforded mainly imidazole $\underline{8}$. Compound $\underline{8}$ presumably formed by intramolecular elimination of CCl₃CH₂OH from amidine intermediate $\underline{7b}$. By thioformylating with excess EtOCSH in liquid H₂S at room temperature under autogenous pressure, $\underline{7c}$ derivatives $\underline{6c}$, $\underline{6d}$, and $\underline{6e}$ were obtained in 75, 89, and 70% yield, respectively: $\underline{6e}$ has mp 92-94°; ir (CCl₄) 3.14, 5.72, 7.00, 8.01, and 9.65 μ ; nmr (CDCl₃) τ 6.20 (s, OCH₃) 4.81 (s, CH₂Ar), 3.96 (d of d, J_{HP} = 22 Hz and J = 8Hz, CHP), 3.11 (d,2, J = 9Hz, ArH), 2.66 (d,2, J = 9Hz, ArH), 0.73 (br m, NH), and 0.50 (d, J = SHz, S=CH); uv (EtOH) 223 (e 13,700) and 261 (e 14,600)m μ ; m/e 375 (M⁺).

No. 46

HCHIN PO(OEt)₂

$$\infty_2$$
R

$$5a$$
, $R = CH_3$
 $5b$, $R = CH_2Ph$
 $5c$, $R = CH_2CCl_3$
 $5d$, $R = CH_2-NO_2$
 $5e$, $R = CH_2-OCH_3$

6a,
$$R = CH_3$$

6b, $R = CH_2Ph$
6c, $R = CH_2CCl_3$
6d, $R = CH_2 - NO_2$
6e, $R = CH_2 - OCH_3$

$$7a$$
, R = CH_2 0CH₃

$$7b$$
, R = CH_2CCl_3

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